Mineralogical characteristics of phlogopite from Dupen kamen, Republic of Macedonia

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Abstract: In this paper are presented mineralogical characteristics of phlogopite from Dupen kamen, Republic of Macedonia. Several samples of phlogopite from Dupen kamenwere analysed by XRD, scanning electron microscope (SEM) and Infrared spectroscopy (IR). The use of these three methods showed that they are very useful methods for rapid mineral analysis contributing important analytical information. The results of the X-ray powder pattern, SEM analysis and Infrared spectroscopyenabled straightforward identification of the studied mineral sample as phlogopite. Phlogopite is located in a series of marblesthat are part of the Pelagonian metamorphic complex. It occurs in yellow and gold yellow monoclinic-prismatic crystals. Transparent to translucent with vitreous lustre. Streak is white. Hardnessis 2-2½ while density is 2.7 - 2.9 g/cm³. Cleavage is perfect on {001}. Concentration of Mg is 16.50-17.98%, Al 6.16-7.31%, Si 20.86-26.84% while K from 8.59 to 9.93%. Phlogpite in marble form Dupen kamen is considered that was formed probably by reaction between dolomite and earlier formed potassium feldspar or muscovite.

Keywords: phlogopite, (SEM)-scanning electron microscope, (X-ray)-powder diffraction(IR) infrared spectroscopy

I. Introduction

Dupen kamen is locatedabout 1 kmnorthwest of the villagePletvarnear Prilep in a series of marbles that are part of the Pelagonian metamorphic complex. Details on the locality Dupen kamen are obtained in the manufacture of base geological map of the R. Macedonia. Although they are regional researches they are a starting point for further research of any kind.

Mineralogical - petrographical studies of the rocks surrounding the village Pletvar, with special deals on marbles are performed by (Marik, L 1940; Dumurdzanov, N., 1976).Marbles of "Dupen kamen" are dolomite and calcite-dolomite. They have granoblastic structure, but in some places structure is a porfiroblastic. They have white or gray colour. White marble are composed of 6% calcite and 94% dolomite. Gray marble are composed of 97% calcite and 3% dolomite after (Rakicevic at all. 1965). Except phlogopite in studied white marblesalso appear following minerals: fluorite, calcite, pyrite, biotite and zoisite. White marbles from Dupen kamen are massive, compact and have high quality.

II. Materials and methods

The phlogopite was identified by Scanning electron microscopy (SEM), coupled with an energy dispersive X-ray spectrometer (EDS), X-ray diffraction (XRD) and Infrared spectroscopy (FTIR). The use of these three methods showed that they are very useful methods for rapid mineral analysis contributing important analytical information.

SEM is especially useful because it gives elemental, mineralogical and morphological data at the same time. Scaning electron microscopy model VEGA3 LMU and EDS-X-act: 10 mm2 connected with INCA 250 EDS software was used.

X-ray diffraction (XRD) is known as the best method for the identification and quantification of minerals. Was used instrument XRD Shimadzu 6100. It was used copper radiation $CuK\alpha = 1.54178$ Å, the voltage of the generator 40 kV, and the current was 30 mA, $2\theta = 2^{\circ}/min$.

Diffuse Reflectance Infrared Fourier Transform Spectroscopy (DRIFTS) was performed using IRPrestige 21 spectrophotometer equipped with a DRS-8000 diffuse reflectance accessory (Shimadzu, Japan). About 1 mg of the powdered sample was diluted to about 5 % by weight in dry KBr. The mixture was then ground to a fine powder and placed into a sample holder (2 mm dia, 1mm deep). The spectra was recorded in the spectral range 400–4000 cm⁻¹ at a resolution of 4 cm⁻¹ with 62 scans. Using this method the infrared spectra was obtained by measuring the diffuse reflected light that was emerged from the powder sample surface after first being absorbed inside the sample and reflected among the particles. For spectra manipulation was used IRSolution 1.5 software (Shimadzu Corporation).

III. Results and discussion

The name phlogopite $KMg_3AlSi_3O_{10}(F,OH)_2$ is from the Greek word meaning for fire, referring to an oft-seen reddish tintn. Appears in prismatic crystals with yellow and gold yellow colour. Cleavage is perfect {001}.Phlogopite is transparent to translucent. Luster is vitreous to pearly, (figure 1). Streak is white. Hardness is 2½, while specific gravity is 2.7 - 2.9 g/cm³. Phlogopite crystalize in monoclinic system 2/m, a = 5.31Å, b = 9.23Å, c = 20.36Å, Z = 4; \beta = 99.3° V = 984.75 Å³.

In thin sections, under microscope phlogopite from Dupen kamen is observed as bronze brown to pale orange crystals with birefringenece of 0.040. Weak pleochroism is observed. It is monoclinic, biaxial (-), although it behaves almost as monoaxial with very small angle between the optical axes $2V = 16^{\circ}$;a=1.553-1.573, b=1.557-1.617, g=1.558-1.618, bire=0.0280-0.0450, Dispersion weak to distinct, r < v. Pleochroisam: (x) pale yellow to orange, (y) brownish yellow, reddish orange.



Figure 1: Phlogopite from Dupen kamen, R. Macedonia

The characterization of phlogopite by use of infrared spectroscopy was subject of study of many authors (Vedder, W. 1964; Farmer, V.C. 1974; Jenkins, D. M. 1989; Šontevska, V. et al. 2008). The FTIR analysis of phlogopite collected from Republic of Macedonia was described in detail by Šontevska et al. (2008). The aim of the infrared analysis included in this study is to demonstrate the DRIFT advantages in the characterization of the phlogopite collected from Dupen kamen locality in R. Macedonia. DRIFT is frequently used technique in the analysis of minerals because it requires simple sample preparation in comparison to other IR techniques (Madejova, J. 2003).

On figure 2 is presented the DR spectrum of the phlogopite from Dupen kamen together with its second derivative profile obtained according to the Savitzky-Golay method using IR Solution software program (Shimadzu Co. Japan). Table 2 summarizes the absorption maxima observed in the spectra of the studied mineral (figure 2) and spectral data from previous analysis of phlogopite mineral.





The absorption maxima in the spectrum obtained by DRIFT spectroscopy in comparison with the spectral findings from previous studies of phlogopite are given in table 1. In the highest spectral region (figure 2) three distinctive peaks arise at 3706, 3667 and 3584 cm⁻¹ corresponding to the stretching vibrations of the OH groups. Such appearance in the DR spectrum suggests that the reflectance has enhanced the peak intensities in comparison with the spectral features in this region obtained using KBr pellet method.

The resolution enhancement of the most intensive band $(1100-900 \text{ cm}^{-1})$ observed in the original spectrum (figure 2) enabled separation of the overlapping bands informative for Si-O-Si stretching vibrations: strong absorptions at 1010 cm⁻¹ and weak absorption at 975 cm⁻¹ observed as a shoulder in the original spectrum. These two bands together with the absorption noticeable at 693 cm⁻¹ attributed to symmetric stretches (table 1) and those observed at the lower frequencies (510 and 480 cm⁻¹) associated to the Si-O-Si bending modes corroborate the spectral findings reported in the literature (Vedder, W. 1964; Farmer, V.C. 1974; Jenkins, D. M. 1989; Šontevska, V. et al. 2008).

According to the previous studies (Jenkins, D. M. 1989) the bands at 813 and 770 cm⁻¹ are linked with the Al-O vibrations whereas the resolved band at 910 and 728 cm⁻¹ could be related to Si-O-Al vibrations. The last band could also be assignable for carbonate matter present in the sample considering the set of bands appearing at: 2525, 1550-1450, 1090, 885 and 728 cm⁻¹ indicative for CO₃ vibrations (Jovanovski, G. et al. 2002).

The second derivative profile of the lower frequency region shows enhanced maximum of the corresponding very weak band at 598 cm⁻¹ observed in the original spectrum. According to Farmer (Farmer, V.C. 1974; Šontevska et al. 2008) this band associated to the OH vibrations might serve as an indication for phlogopite low OH molar absorption coefficient.

Moreover, the spectral features in the 2920-2850 cm⁻¹ region (figure 2) indicate presence of some organic matter in the studied mineral. The appearance of absorption band centered at around 3400 cm⁻¹ and the weak band at around 1640 cm⁻¹ can be explained by the H-O-H moleculesstretches of adsorbed water.

This study natural phlogopite	Jenkins, D. M, (1989) synthetic-powder		Šontevska, V et al (2008) natural phlogopite		
3706 w			3705 vw	v(OH)	
3667 w					
3584 w					
1010 vs	995 vs	Si-O in plane	1011 vs	v(Si-O-Si)	
975sh*	960 vs	Si-O in plane	977 sh, s	v (Si-O-Si)	
910*	915 s*	Si-O-Al in plane			
813 m	822 ms	Al-O perp	813 w	v (Al-O-Al)	
770 vw	760 vw	Al-O in plane			
728 w	725 w	Al-O-Si in plane	727 sh	v (Al-O-Si)	
693 m	690 ms	Si-O-Mg	692 m	v _s (Si-O-Si)	
652 vw	655 w	Al-O in plane		v(Al-O)	
598vw	592 ms	OH libration		L(OH)	
510 sh	520 vs*	Si-O perp	511 sh	δ(Si-O-Si)	
480 s	495 vs	Mg-O	470 vs	δ(Si-O-Si)	
	460 vs	Si-O in plane			

Table 1:The absorption maxima in the spectrum obtained by DRIFT spectroscopy in comparison with the spectral findings from previous studies of phlogopite.

Abbreviation: w-weak, v-very, s-strong, m-moderately, sh-shoulder; * = resolved band. In Jenkins, D. M. 1989: orientation of vibration – parallel (in plane) or perpendicular (perp) to the cleavage plane.

Morphology of phlogopite from Dupen kamen observed with SEM shows thin platy, layered crystals (figure3). Ideal chemical formula of phlogopite is $KMg_3(Si_3Al)O_{10}(F,OH)_2$. The Mg content in the studied phlogopite is 16.50-17.98%, Al is between 6.16-7.31%, Si is from 20.86 to 26.84% while K from 8.59 to 9.93%.

According to its chemical composition and considering the Nomenclature of the micas (Rieder et al. 1998) the mineral is determined as phlogopite. On table 2 is given the chemistry of phlogopite from Dupen kamen.



	Spectrum 1		Spectrum 2		Spectrum 3	
Element	Weight%	Atomic%	Weight%	Atomic%	Weight%	Atomic%
0	44.43	58.30	39.09	52.88	46.74	60.44
Mg	17.64	15.23	17.98	16.01	16.50	14.04
Al	6.29	4.89	6.16	4.94	7.31	5.61
Si	21.79	16.29	26.84	20.68	20.86	15.37
K	9.85	5.29	9.93	5.49	8.59	4.55
Totals	100.00	100.00	100.00	100.00	100.00	100.00

Table 2: Chemistry of phlogopite from Dupen kamen (spectrum 1, 2 and 3)

The first researcher to describe X-ray diffraction examination of mica was Mauguit (1927,1928), who determined the symmetry, unit cell dimensions, and cell content of several natural specimens. Shortly thereafter the basic crystal structure of mica was proposed almost simultaneously by Pauling (1930) and Jackson and West (1930, 1933). Pabst (1955) presented theoretical atomic coordinates for an "ideal" C2f m, one-layer, trioctahedral mica and this model has been subsequently described and illustrated by several authors (e.g., Bragg et al 1965, p. 254 et seq). Phlogopite structure is basic for 1M trioctahedral micas. Rayner (1974) refined this structure using single-crystal neutron diffraction reflection data, which permitted precise determination of H atom position.

The most intense registered maxima in the studied powder diagram (figure 4) are: 3.338(100), 2.614(30), 2.009(30), 3.390(20), 2.513(16), 2.429(16). They were compared with the corresponding maxima in the diagram of phlogopite sampleJCPDScards 01 073 7742. The comparison has shown that the X-ray powder pattern of the natural phlogopite taken from the JCPDS card are practically identical with the obtained results in this stady.



Figure 4:X-ray diagram of phlogopite

IV. Conclusion

After summarizing the data collected in this research, we canconfirm that the studied mineral sample is phlogopite. The straightforward identification of the studied mineral sample was enable by X-ray powder pattern, SEM analysis and IR spectroscopy.

Phlogopite is located in a series of marbles that are part of the Pelagonian metamorphic complex. It occurs in yellow and gold yellow monoclinic-prismatic crystals. Streak is white. Phlogopite is transparent to translucent. Lustre is vitreous. Hardnes is $2-2\frac{1}{2}$ and the density 2.7 - 2.9 g/cm³. Cleavage is perfect on {001}.

The Mg content in the studied phlogopite varies from 16.50 to17.98%, Al is within between6.16 and 7.31%, Si vary from 20.86-26.84% while concentrations of K in range of 8.59 to 9.93%. Resultsobtained byX-rayinvestigations in goodagreement with JCPDS cards 01 073 7742.

The spectral features of phlogopite achieved by use of DRIFT spectroscopy are in agreement with the previous studies of this mineral by use of KBr pellet method.

Phlogpite in marble form Dupen kamen is considered that was formed probably by reaction between dolomite and earlier formed potassium feldspar or muscovite.

 $3CaMg(CO_3)_2 + KAlSi_3O_8 + H_2O \rightarrow (OH)_2 KMg_3 Al Si_3O_{10} + 3CaCO_3 + 3CO_2$ $3CaMg(CO_3)_2 + (OH)KAlSi_3O_8 \rightarrow (OH)_2 KMg_3 Al Si_3O_{10} + 3CaCO_3 + 3CO_2 + Al_2O_3$

Phlogopite is economically important invarious products of our daily life.

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